

**3-(4-Bromophenyl)-1-(4-nitrophenyl)-
prop-2-en-1-one**

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Key indicators

Single-crystal X-ray study
 $T = 100\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$
 R factor = 0.032
 wR factor = 0.085
Data-to-parameter ratio = 30.7

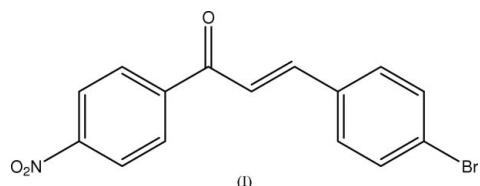
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound, $\text{C}_{15}\text{H}_{10}\text{BrNO}_3$, the molecules are arranged into infinite chains through $\text{Br}\cdots\text{O}$ short interactions. The chains are stacked to form layers. These layers are interconnected by $\text{C}-\text{H}\cdots\text{O}$ interactions. There are two molecules in the asymmetric unit.

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Comment

Chalcones have been claimed to be a class of compounds that play a vital role in antitumour (Mishra *et al.*, 2001), anti-inflammatory (Ko *et al.*, 2003; Tuchinda *et al.*, 2002) and antimarial (Domínguez *et al.*, 2001) activities. It has also been documented that chalcones exhibit extremely high and fast non-linearity (Fichou *et al.*, 1988; Zhang *et al.*, 1990; Zhao *et al.*, 2000) and are easy to crystallize as non-centrosymmetric structures. Another importance of this type of compounds is their high photosensitivity and thermal stability, which are used in developing various crystalline electro-optical devices (Williams *et al.*, 1983; Chemla *et al.*, 1987). In view of their exceptional behaviour, chalcones have been the subject of several experimental and theoretical studies, aimed mainly at determining their crystal structures (Moorthi *et al.*, 2005; Radha Krishna *et al.*, 2005; Uchida *et al.*, 1995). In this work, we report the synthesis and crystal structure of 3-(4-bromophenyl)-1-(4-nitrophenyl)prop-2-en-1-one, (I).



There are two molecules in the asymmetric unit of (I). The bond parameters are comparable with those in related structures (Teh *et al.*, 2006; Ng *et al.*, 2006; Patil *et al.*, 2006). The dihedral angle between the benzene rings is $12.83(7)^\circ$ for molecule *A* and $41.15(7)^\circ$ for molecule *B*. The nitro group attached at C13 is twisted away from the C10–C15 benzene ring, with torsion angles O2–N1–C13–C14 = $-5.8(2)$ and $12.2(2)^\circ$, and O3–N1–C13–C12 = $-6.0(2)$ and $12.9(2)^\circ$ for molecules *A* and *B*, respectively. The least-squares plane through the enone fragment makes dihedral angles of $16.36(6)$ and $17.09(6)^\circ$ with the C1–C6 and C10–C15 benzene rings for molecule *A*, and of $24.49(11)$ and $18.56(12)^\circ$ for molecule *B*.

In the crystal structure of (I), the intramolecular C7–H7A···O1 interaction generates an *S*(5) ring motif (Bernstein *et al.*, 1995) for both molecules *A* and *B*. The molecules are

arranged into infinite chains through short $\text{Br}1\text{A}\cdots\text{O}2\text{B}(-1+x, 1+y, z)$ [3.0417 (14) Å] interactions and also weak short $\text{Br}1\text{A}\cdots\text{O}3\text{B}(x, 1+y, z)$ [3.3741 (13) Å] interactions. These chains are stacked to form layers which are interconnected by C—H···O interactions (Table 1).

Experimental

Chalcone derivative (**I**) was obtained by the condensation of 4-bromobenzaldehyde (0.01 mol) and 4'-nitroacetophenone (0.01 mol) in ethanol (60 ml) in the presence of NaOH (2 ml, 30%). The resulting crude solid compound was collected by filtration, dried and recrystallized from acetone. Crystals suitable for X-ray diffraction study were grown by slow evaporation of an acetone solution over a period of 7 d.

Crystal data

$C_{15}H_{10}BrNO_3$	$Z = 4$
$M_r = 332.15$	$D_x = 1.740 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 5.9341 (1) \text{ \AA}$	Cell parameters from 6533 reflections
$b = 7.9193 (1) \text{ \AA}$	$\theta = 0.8\text{--}35.0^\circ$
$c = 27.2996 (4) \text{ \AA}$	$\mu = 3.25 \text{ mm}^{-1}$
$\alpha = 89.938 (1)^\circ$	$T = 100.0 (1) \text{ K}$
$\beta = 85.622 (1)^\circ$	Block, yellow
$\gamma = 82.357 (1)^\circ$	$0.55 \times 0.39 \times 0.25 \text{ mm}$
$V = 1267.75 (3) \text{ \AA}^3$	

Data collection

Bruker SMART APEX2 CCD area-detector diffractometer
 ω scans
Absorption correction: multi-scan
(*SADABS*; Bruker, 2005)
 $T_{\min} = 0.286$, $T_{\max} = 0.444$
41078 measured reflections

11069 independent reflections
9020 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.036$
 $\theta_{\text{max}} = 35.0^\circ$
 $h = -9 \rightarrow 9$
 $k = -12 \rightarrow 12$
 $l = -44 \rightarrow 39$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.033$
 $wR(F^2) = 0.085$
 $S = 1.11$
11069 reflections
361 parameters
H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0409P)^2 + 0.3692P]$$

where $P = (F_o^2 + 2F_c^2)/3$

 $(\Delta/\sigma)_{\text{max}} = 0.002$
 $\Delta\rho_{\text{max}} = 0.94 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.57 \text{ e \AA}^{-3}$

Table 1
Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$C7\text{A}-\text{H}7\text{AA}\cdots\text{O}1\text{A}$	0.93	2.46	2.804 (2)	102
$C7\text{B}-\text{H}7\text{BA}\cdots\text{O}1\text{B}$	0.93	2.49	2.812 (2)	100
$C15\text{B}-\text{H}15\text{B}\cdots\text{O}1\text{A}^{\text{i}}$	0.93	2.58	3.240 (2)	129

Symmetry code: (i) $-x + 2, -y, -z + 1$.

H atoms were placed in calculated positions, with C—H distances of 0.93 Å. The $U_{\text{iso}}(\text{H})$ values were constrained to be $1.2U_{\text{eq}}$ (carrier) for all H atoms.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *APEX2*; data reduction: *SAINT* (Bruker, 2005); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1998); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used

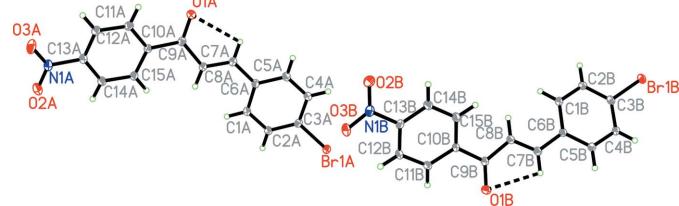


Figure 1

The asymmetric unit of (**I**), showing 50% probability displacement ellipsoids and the atomic numbering. Dashed lines indicate hydrogen bonds.

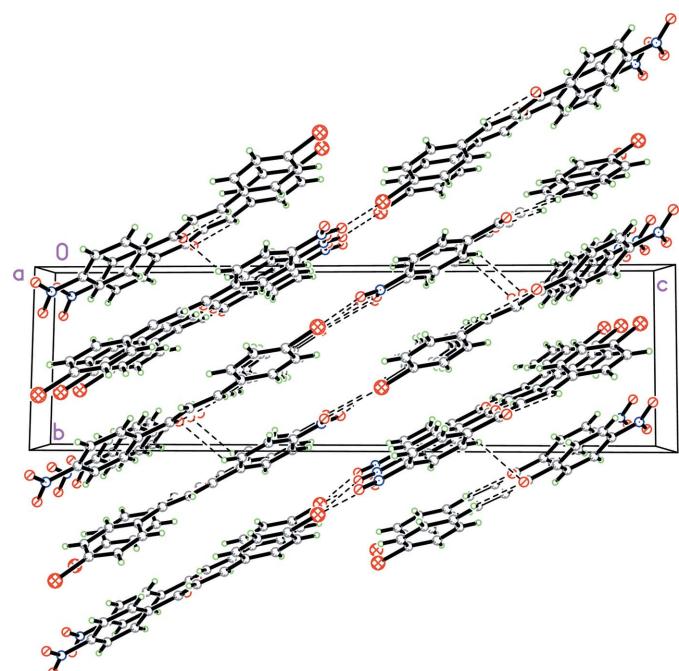


Figure 2

The crystal packing of (**I**), viewed down the a axis. Hydrogen bonds are shown as dashed lines.

to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2003).

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